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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/645,690	08/24/2000	Lizhong Sun	4215/PDD/CMP/RKK	4428
7	590 01/02/2004		EXAM	INER
PATENT COUNSEL		WINTER, GENTLE E		
APPLIED MA	TERIALS, INC			
P.O. BOX 450A			ART UNIT	PAPER NUMBER
SANTA CLAR	A. CA 95052		1746	

DATE MAILED: 01/02/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

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1	Application No.	Applicant(s)				
Office Action Summary	09/645,690	SUN ET AL.				
Onice Action Summary	Examiner	Art Unit				
The MAIL INC DATE of this communication and	Gentle E. Winter	1746				
The MAILING DATE of this communication app Period for Reply	Jears on the cover sheet with the C	correspondence address				
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION.  Extensions of time may be available under the provisions of 37 CFR 1.1: after SIX (6) MONTHS from the mailing date of this communication.  If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period v Failure to reply within the set or extended period for reply will, by statute.  Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).  Status	36(a). In no event, however, may a reply be tir y within the statutory minimum of thirty (30) day will apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	nely filed  rs will be considered timely.  the mailing date of this communication.  10 (35 U.S.C. § 133).				
1) Responsive to communication(s) filed on 03 No	ovember 2003.					
2a)⊠ This action is <b>FINAL</b> . 2b)☐ This	action is non-final.					
<ol> <li>Since this application is in condition for allower closed in accordance with the practice under E</li> </ol>	nce except for formal matters, pro Ex parte Quayle, 1935 C.D. 11, 45	osecution as to the merits is 53 O.G. 213.				
Disposition of Claims						
4a) Of the above claim(s) is/are withdraw 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-18,26-31 and 33 is/are rejected. 7) ☐ Claim(s) is/are objected to.	Claim(s) <u>1-18,26-31 and 33</u> is/are rejected.					
Application Papers	'					
9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) accomposed and accomposed and any objection to the composed and accomposed accomposed accomposed and accomposed	epted or b) objected to by the I drawing(s) be held in abeyance. Ser ion is required if the drawing(s) is ob	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).				
Priority under 35 U.S.C. §§ 119 and 120						
12) Acknowledgment is made of a claim for foreign  a) All b) Some * c) None of:  1. Certified copies of the priority documents  2. Certified copies of the priority documents  3. Copies of the certified copies of the prior  application from the International Bureau  * See the attached detailed Office action for a list that the companies of a claim for domestic since a specific reference was included in the first specific reference was included in the first specific reference was included in the first sentence of the companies of the foreign language process.	s have been received. s have been received in Applicative documents have been received in (PCT Rule 17.2(a)). of the certified copies not received priority under 35 U.S.C. § 119(a) at sentence of the specification or visional application has been received priority under 35 U.S.C. §§ 120	on No  and in this National Stage  and.  but the interior of the inter				
Attachment(s)						
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449) Paper No(s)	<ol> <li>Notice of Informal P</li> </ol>	(PTO-413) Paper No(s) atent Application (PTO-152)				
S Patent and Trademark Office						

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#### DETAILED ACTION

## Response to Arguments

## 1. Applicants argued:

Applicants respectfully disagree with the Examiner and point out that Small et al. does not teach, show, or suggest a method of cleaning a polishing pad surface comprising applying to the polishing pad surface a cleaning composition including about 0.1 to about 3.0 M.% of at least one organic compound containing one or more amine or amide groups, an acid or a base in an amount such that the composition has a PH of about 5.0 to about 12.0, and water, as recited in claims 1, 12, 26, and claims dependent therefrom.

2. Applicant's argument is accepted. Small is provided for the teaching of cleaning a wafer using a polishing pad, and the secondary reference is provided for the teaching that the chemistry used to clean the pad is effective on the wafer. As was indicated preciously, parsing the art and arguing that the references, taken individually do not amount to an anticipation rejection is unlikely to result in the indication of allowable subject matter.

# 3. Applicants argued:

In addition, the Examiner has erred in stating that explicitly disclosed range of the amine is 3-20%. Applicants respectfully point out that the cited 3-20% of amine concentration in water is not directed to the post clean treatment composition of Small as the Examiner stated, but rather a statement of the background problem that need to be solved by Small's post clean treatment solution so as to rapidly neutralize amine and prevent corrosion of metal structure and other features on a wafer. (See, column 3, lines 54-62)

4. Applicants appear to acknowledge that the prior art of record discloses that the disclosed amine range is desirable, that is 3% and less. The art recognizes that significant quantities of amine provide deleterious results.

## Applicant continued:

Further, with regard to the PH range, the acidic pH range of about 3.5 to 7 of Small's post clean treatment composition is provided to neutralize the pH of a wafer until the pH reaches 7 and prevent corrosion of metal structure on a wafer during rinsing.

6. Exactly what is intended by a step is less relevant than that the same step is undertaken, a point seemingly acknowledged by applicant.

# Applicant further argued:

Small does not teach, show, or suggest the claimed composition in the claimed range for the claimed method and nowhere does Small provide motivation to use Small's composition for cleaning a polishing pad between polishing a first and a second wafer. The advantage of Small's post clean treatment composition as a rinsing composition for a wafer as stated by the Examiner and disclosed in Small does not provide motivation or rationale for cleaning a polishing pad between polishing a first and a second wafers. Withdrawal of the rejection is respectfully requested.

7. At this time, it is not clear if applicant is suggesting that the motivation for cleaning the polishing pad exists, or if the motivation for the disclosed method exists. It is presumed that applicant is not suggesting that there is no motivation for cleaning the pad. This examiner relies on the motivation and rationale of record.

## Applicant concluded:

Applicant respectfully traverses the rejection. Small Svirchevski et el., and Kennedy et al. have been discussed above. Small does not teach, show, or suggest the method and composition as claimed and Svirchevski et al. and Kennedy et al. do not teach, show, or suggest the composition as claimed. As discussed above, there is no motivation in the references to combine Small's wafer cleaning composition with Svirchevski et al. and Kennedy et al.'s polishing pad cleaning method despite the disclosure of flow rates in the method of Svirchevski et al. and Kennedy et al. Further, Small's composition works by rinsing a wafer in a bath such as immersing the wafer for a duration of 30 min (Example 2) to 24 hours (Example 3) rather than for cleaning a polishing pad surface for a short time period after CMP. Thus, Small does not teach, show, or suggest applying Small's post clean treatment solution to a polishing pad for about 3 seconds to about 20 seconds after conducting CMP as recited in claims 18, and 30, and can not be served as basis for the Small composition to be combined with Svirchevski et al. and Kennedy et al.

8. Since applicant does not argue that the aggregated references fail to teach the invention, and since the rejection is base on obviousness, applicant's arguments drawn to arguments that the references individually do not teach the invention are not argued. This examiner does not recognize the above as an attack on the motivation however the same is presumed. This

examiner believes that the reasons for making the claimed motivation are sound and proper. As such, the rejections cannot be withdrawn at this time.

#### Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 1. Claims 1-3, 8, 12-14, 16, 26, 27 and 33 rejected under 35 U.S.C. 103(a) as being unpatentable over United States Patent No. 5,981,454 to Small and United States Patent No. 6,352,595 to Syirchevski et al.
- 2. Small discloses a method of cleaning a wafer using a polishing pad see e.g. column 2, line 55. Small discloses applying to the wafer a cleaning composition comprising a composition that has an amine concentration that includes at least one endpoint of the claimed range, specifically 3.0 Wt% of at least one organic compound containing one or more amine or amide groups. It is noted that the solution is applied to the polishing pad surface by transfer from the wafer of Small. Small discloses this at e.g. column 2, line 55 et seq. The explicitly disclosed range is 3-20% (column 3, line 52 et seq.) however figure 3 appears to contemplate a range of activity below 3% (see e.g. column 4, line 6 et seq. and also see figure 3). Small further discloses an acid or base such that the composition has a pH of between 3.5 and 7, which anticipates the range 5.0 to about 12.0, and water. See (column 2, line 37 et seq. and column 3, line 52 et seq.). Small further discloses using DI water, ethylene diamine, and acetic acid

(column 14, line 43 through column 16). A water rinse is also disclosed (column 1, line 38 et seq.). It is noted that the argument that the Small reference is drawn applying the cleaning solution to the pad and not the wafer, is not persuasive because applying the solution to the wafer is deemed an intermediate step. In the end, the solution is applied to the cleaning pad. If applicant wants to recite that the cleaning solution is applied directly to the pad, such would seemingly distinguish the instant claims over Small. However, it follows that a solution that is effective for removing compounds from the wafer would similarly be effective for removing compounds from a pad.

- 3. With respect to claim 12 and dependant claims, the arguments and claim recitation of a *first* and *second* wafer and an intermediate cleaning step precludes the possibility of a wafer being present during the cleaning and as such the anticipation rejection is withdrawn.
- 4. The secondary reference provides the missing element and Small explicitly provides the motivation for making the instant combination. Specifically, Svirchevski discloses a method of cleaning a chemical mechanical polishing (CMP) pad that has already been used for performing a CMP operation on a wafer surface, the CMP pad having a residue on a surface of the CMP pad, the method includes applying chemicals to the surface of the CMP pad and thereafter rinsing the pad. The chemical disclosed in Small is disclosed to be useful because it eliminates the need for flammable solvents, lowers transition metal ion concentrations, and has a high neutralization capacity. The rational for cleaning between a first and second wafer would include minimization of contamination build-up and minimization of subsequent bath contamination.
- Claims 12-14, and 16 are rejected under 35 U.S.C. 103(a) as being unpatentable over
   Small as set forth above and United States Patent No. 6,352,595 to Svirchevski et al.

6. Claims 12-14 and 16 disclosing the sequential steps of conducting a CMP on a first wafer surface of a first water containing copper or copper based alloy on a surface-polishing pad. It is not altogether clear if "water" was intended to be "wafer". Thereafter the first wafer is removed from the pad, and cleaning composition is applied to the pad, followed by a rinsing step. Finally, a second wafer is provided and the above steps are repeated. The composition is identically disclosed in Small as set forth above. What is not explicitly disclosed is the cleaning of the pad between polishing a first wafer and second wafer. This sequential order appears to require that a wafer is not take part of the pad cleaning operation. This inference is supported by the terms "sequential" and the identification of a "first" and a "second" wafer, which are not in contact with the pad during the cleaning operation. Because the each and every limitation of the claim is not identically disclosed, and is apparently not inherent the anticipation rejection is withdrawn at this time. The secondary reference provides the missing element and Small explicitly provides the motivation for making the instant combination. Specifically, Svirchevski discloses a method of cleaning a chemical mechanical polishing (CMP) pad that has already been used for performing a CMP operation on a wafer surface, the CMP pad having a residue on a surface of the CMP pad, the method includes applying chemicals to the surface of the CMP pad and thereafter rinsing the pad. The chemical disclosed in Small is disclosed to be useful because it eliminates the need for flammable solvents, lowers transition metal ion concentrations, and has a high neutralization capacity. The rational for cleaning between a first and second wafer would include minimization of contamination build-up and minimization of subsequent bath contamination.

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- 7. Claims 4-7, 9-11, and 28-31, as currently understood, are rejected under 35 U.S.C. 103(a) as being unpatentable over Small and United States Patent No. 6,280,299 to Kennedy.
- 8. Each and every limitation of claims 6, 7, 9, 10, and 11 are identically disclosed in Small, except that Small apparently fails to explicitly disclose applying the solution to a rotating polishing pad at a flow rate of about 10 to 600 ml/min. Kennedy et al. discloses using a flowrate between 230 and 6000 ml/min (e.g. column 6, line 58 et seq.). The artisan would have been motivated to make the instant combination for the reasons explicitly set forth in Kennedy et al. Kennedy discloses that the pad cleaning flowrates and pressures are optimized based on the conditions and materials used in the pad cleaning process. In a larger sense, the artisan would have been motivated to select a flow rate high enough to reduce the pad loading to an acceptable level, while minimizing solvent waste. Similarly, it is submitted that duration of the flow would be a matter of routine optimization, but is explicitly disclosed in Kennedy et al. as about 5 to 20 seconds (see e.g. column 7, line 47 et seq.). Again, the motivation is explicitly disclosed in Kennedy et al. and the instant invention appear to be performing substantially the same task, in substantially the same way, for substantially the same reason.
- 9. With specific respect to claims 4 and 28 the claims, as amended disclose a pH range of "about 8 to about 11". While these values may not be contextually taught with identical values the taught value of "about 7" seemingly would read on "about 8", however if it is asserted that there are real material differences observed between a pH of 7 and a pH of 8 seemingly such evidence would overcome the obvious rejection. But note that pH values of greater than 10 (and therefore including 11) are disclosed in Small, see e.g. column 4, line 3.

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- Claims 15, 17, and 18 rejected under 35 U.S.C. 103(a) as being unpatentable over Small Svirchevski and Kennedy as set forth above.
- 11. Each and every limitation of claims 17, and 18 is disclosed in the combination of Small and Svirchevski as set forth above with respect to claim 12, except that Small and Svirchevski apparently fail to explicitly disclose applying the solution to a rotating polishing pad at a flow rate of about 10 to 600 ml/min. Kennedy et al. discloses using a flowrate between 230 and 6000 ml/min (e.g. column 6, line 58 et seq.). The artisan would have been motivated to make the instant combination for the reasons explicitly set forth in Kennedy et al. Kennedy discloses that the pad cleaning flowrates and pressures are optimized based on the conditions and materials used in the pad cleaning process. In a larger sense, the artisan would have been motivated to select a flow rate high enough to reduce the pad loading to an acceptable level, while minimizing solvent waste. Similarly, it is submitted that duration of the flow would be a matter of routine optimization, but is explicitly disclosed in Kennedy et al. as about 5 to 20 seconds (see e.g. column 7, line 47 et seq.). Again the motivation is explicitly disclosed in Kennedy et al. specifically, optimizing the cleaning of the pad. Further, Kennedy et al. and the instant invention appear to be performing substantially the same task, in substantially the same way, for substantially the same reason.
- 12. With specific respect to claim 15 the claim, as amended discloses a pH range of "about 8 to about 11". While these values may not be contextually taught with identical values the taught value of "about 7" seemingly would read on "about 8", however if it is asserted that there are real material differences observed between a pH of 7 and a pH of 8 seemingly such evidence

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would overcome the obvious rejection. But note that pH values of greater than 10 (and therefore including 11) are disclosed in Small, see e.g. column 4, line 3.

#### Conclusion

13. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

14. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Gentle E. Winter whose telephone number is (703) 305-3403. The examiner can normally be reached on Monday-Friday, 8:30-5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Randy P. Gulakowski can be reached on (703) 308-4333. The fax phone numbers for

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the organization where this application or proceeding is assigned are (703) 872-9310 for regular communications and (703) 872-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703) 308-0661.

Gentle E. Winter Examiner Art Unit 1746

December 22, 2003

Teinal Elanini

ZEINAB EL-ARINI PRIMARY EXAMINER